# PYRUVATE SYNTHESIS FROM ACETYL COENZYME A AND CARBON DIOXIDE WITH NADH<sub>2</sub> OR NADPH<sub>2</sub> AS ELECTRON DONORS

## G. GOTTSCHALK and A. A. CHOWDHURY

Institute of Microbiology, University of Göttingen, 34 Göttingen, Germany

Received 4 February 1969

#### 1. Introduction

In Clostridium kluvveri reduced ferredoxin is acting as electron donor in two reactions: the evolution of hydrogen gas during the ethanol-acetate fermentation [1] and the synthesis of pyruvate from acetyl-CoA CoA and CO<sub>2</sub> [2,3]. The first reaction is important because it enables C. kluyveri to gain ATP by substrate phosphorylation [4]. The second reaction provides the cell with pyruvate needed for the biosynthesis of cellular constituents [5]. The physiological reaction, however, by which Fd<sup>+</sup> is reduced and regenerated for the above processes is not known. In this paper we report that a cell-free system of C. kluvveri supplemented with NADH2 or NADPH2 and acetyl-CoA catalyzes the reduced Fd dependent synthesis of pyruvate in a nitrogen atmosphere. With a cell-free system of C. pasteurianum a similar reaction was not observed.

### 2. Materials and methods

C. kluyveri was grown and cell-free extracts were prepared as described previously [6]. C. pasteurianum was cultivated in the medium of ref. [7] and Fd was purified according to ref. [8]. The enzymic reactions were carried out in Warburg vessels containing Sephadex G-25 treated cell-free extract in 50 mM potassium phosphate buffer, pH 7.0, and 25 mM 2-mercapto-

ethanol in sidearm 1, KH<sup>14</sup>CO<sub>3</sub> solution in sidearm 2 and the other reactants in the main compartment. The Warburg vessels were flushed with purified nitrogen gas for 10 min. Following a preincubation period of 20 min, the reactions were run for 20 min at 30°. After stopping with perchloric acid and removal of the <sup>14</sup>CO<sub>2</sub> the remaining non-volatile radioactivity was measured in a Packard Tri-Carb using Bray's [9] scintillation fluid. Paper chromatography was carried out according to ref. [10]. LDH (360 U/mg) was purchased from Boehringer, Mannheim.

## 3. Results

As is well known [2,3] cell-free extracts of *C. kluy-veri* catalyzed the synthesis of pyruvate from acetyl-CoA and CO<sub>2</sub> in a hydrogen atmosphere (table 1). The rate of pyruvate formation increased when LDH and NADH<sub>2</sub> were added as trapping reagents. An incorporation of <sup>14</sup>CO<sub>2</sub> into non-volatile compounds was also observed when the reaction was carried out under nitrogen in the presence of relatively high concentrations of NADH<sub>2</sub> or NADPH<sub>2</sub>. Pyruvate trapping reagents such as LDH or semicarbazide increased the rate of <sup>14</sup>CO<sub>2</sub> fixation. Similar experiments were performed with cell-free extracts of *C. pasteurianum*. Although the pyruvate synthase system was active in these extracts no <sup>14</sup>CO<sub>2</sub> was incorporated with NADH<sub>2</sub> or NADPH<sub>2</sub> as terminal electron donors.

Reactions 5 and 7 of table 1 were run with <sup>14</sup>CO<sub>2</sub> of a higher specific radioactivity and the products formed were identified. Chromatography and cochro-

<sup>\*</sup> Abbreviations used: ferredoxin = Fd; lactate dehydrogenase = LDH.

Table 1

Reductive carboxylation of acetyl-CoA by cell-free extracts of C. kluyveri and of C. pasteurianum with hydrogen or NAD(P)H<sub>2</sub> as terminal electron donors.

| Components added or omitted                        | Atmo-<br>sphere | <sup>14</sup> CO <sub>2</sub> incorporated |                  |                 |                  |
|--|-----------------|--|------------------|-----------------|------------------|
|  |                 | C. kluyveri                                |                  | C. pasteurianum |                  |
|  |                 | (cpm/assay)                                | (U/g of protein) | (cpm/assay)     | (U/g of protein) |
| 1. –   | Н               | 24,460                                     | 1.42             | ~               | _                |
| 2. + NADH <sub>2</sub> ; + LDH                     | Н               | 144,130                                    | 8.38             | 110,200         | 9.75             |
| 3. =   | N               | 90   | _                | 116             | _                |
| 4. + NADH <sub>2</sub>                             | N               | 3,650                                      | 0.21             | 166             |                  |
| 5. + NADH <sub>2</sub> ; + LDH                     | N               | 57,170                                     | 3.32             | 180             | _                |
| 6. + NADH <sub>2</sub> ; + LDH; – acetyl phosphate | N               | 140  | _                | 120             | _                |
| 7. + NADPH <sub>2</sub>                            | N               | 5,790                                      | 0.34             | 284             |                  |
| 8. + NADPH <sub>2</sub> ; + semicarbazide          | N               | 9,550                                      | 0.55             | 196             | -                |

The assay system contained in a total volume of 2.0 ml: potassium phosphate buffer, pH 7.0, 50 mM; 2-mercaptoethanol, 25 mM; potassium lithium acetyl phosphate, 25 mM; KH $^{14}$ CO $_{3}$  (279,000 cpm/ $\mu$ mole), 10 mM; coenzyme A, 0.5 mM; and 0.4 ml of Sephadex G-25 treated extract of C. kluyveri (7.7 mg of protein/ml) and of C. pasteurianum (5 mg of protein/ml), respectively. NADH $_{2}$ , 2.5 mM; NADPH $_{2}$ , 5 mM; LDH, 100  $\mu$ g; and semicarbazide, 10 mM were added as indicated. The experimental conditions were described in Methods. U =  $\mu$ moles/min.

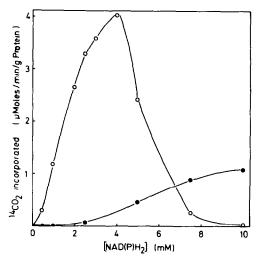
Table 2

Effect of ferredoxin on the reductive carboxylation of acetylCoA with NADH<sub>2</sub> as terminal electron donor.

| Ferredoxin added | <sup>14</sup> CO <sub>2</sub> incorporated |                  |  |  |
|------------------|--|------------------|--|--|
| (μg/assay)       | (cpm/assay)                                | (U/g of protein) |  |  |
| 0                | 80   |                  |  |  |
| 10               | 140  | _                |  |  |
| 25               | 4,230                                      | 0.52             |  |  |
| 50               | 9,970                                      | 1.23             |  |  |

The assay system contained the components described in table 1. NADH<sub>2</sub>, 5 mM; LDH, 100  $\mu$ g; and cell-free extract (3.6 mg protein/ml), 0.4 ml were added. To remove ferredoxin the Sephadex G-25 treated extract was passed through a DEAE-cellulose column (3.0  $\times$  0.8 cm) equilibrated against 50 mM potassium phosphate buffer, pH 7.0.

matography with authentic samples in two solvent systems revealed the formation of pyruvate (74%) and alanine (26%) in reaction 7. Reaction 5 yielded lactate as the only radioactive product. It was identified by chromatography in 1-propanol: ammonia and isolated by chromatography of the reaction mixture on Dowex-1-formate. Degradation of the <sup>14</sup>C-lactate by manganese dioxide [11] revealed that the radioactivity was present exclusively in the carboxyl group.



It is apparent from table 2 that the reductive carboxylation of acetyl-CoA with NADH<sub>2</sub> as electron donor was dependent on the presence of Fd. Extracts

passed through a DEAE-cellulose column catalyzed the reaction only after the addition of Fd.

The dependence of the amount of <sup>14</sup>CO<sub>2</sub> fixed on the concentration of NADH<sub>2</sub> and NADPH<sub>2</sub> is shown in fig. 1. It can be seen that high concentrations of NADH<sub>2</sub> strongly inhibit the reaction. At concentrations up to 5 mM, NADH<sub>2</sub> is more effective as an electron donor than NADPH<sub>2</sub>.

### 4. Discussion

Cell-free extracts of *C. kluyveri* catalyze the synthesis of pyruvate from acetyl-CoA, CO<sub>2</sub> and NADH<sub>2</sub> or NADPH<sub>2</sub>. The reaction is not catalyzed by extracts of *C. pasteurianum* and *Chromatium* as was shown by Buchanan et al. [12]. The enzyme activity responsible for the reduction of Fd is therefore not necessarily present in all anaerobes containing NAD or NADP reductase.

## Acknowledgements

The authors are grateful to Miss S.Dittbrenner for

excellent assistance. This work was supported by the Deutsche Forschungsgemeinschaft and by the Fonds der Chemischen Industrie.

## References

- [1] W.W.Fredricks and E.R.Stadtman, J. Biol. Chem. 240 (1965) 4065.
- [2] I.G.Andrews and J.G.Morris, Biochim. Biophys. Acta 97 (1965) 176.
- [3] J.R.Stern, in: Non-Heme Iron Proteins: Role in Energy Conversion, ed. A.San Pietro (Antioch, Yellow Springs, Ohio) p. 199.
- [4] R.K.Thauer, K.Jungermann, H.Henninger, J.Wenning and K.Decker, European J. Biochem. 4 (1968) 173.
- [5] N.Tomlinson, J. Biol. Chem. 209 (1954) 597.
- [6] G.Gottschalk and H.A.Barker, Biochemistry 5 (1966) 1125.
- [7] J.E.Carnahan and J.E.Castle, J. Bacteriol. 75 (1958)
- [8] L.E.Mortensen, Biochim. Biophys. Acta 81 (1964) 71.
- [9] G.A.Bray, Anal. Biochem. 1 (1960) 279.
- [10] P.Hirsch, Arch. Mikrobiol. 46 (1963) 53.
- [11] W.Sakami, Handbook of Isotope Tracer Methods (School of Medecine, Ohio, 1955) p. 46.
- [12] B.B.Buchanan, R.Bachofen and D.I.Arnon, Proc. Natl. Acad. Sci. 52 (1964) 839.